

4 α ,15-DIHYDROENCELIN AND RELATED SESQUITERPENE ACIDS FROM *PERYMENIUM FEATHERSTONEI*

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Key Word Index—*Perymenium featherstonei*; Compositae; sesquiterpenes; eudesmanolide; eudesmanoic acids.

Abstract—*Perymenium featherstonei* afforded, in addition to the known *ent*-kaurene derivative 4 α ,15-dihydroencelin, two closely related epimeric acids.

INTRODUCTION

So far only one species of the South and Central American genus *Perymenium* has been studied chemically [1]. As only *ent*-kaurene derivatives were isolated a further species, *P. featherstonei* Blake, collected in Peru, was investigated to see whether chemotaxonomic relationships to other genera of the subtribe Ecliptinae [2] exist. The results will be discussed in this paper.

RESULTS AND DISCUSSION

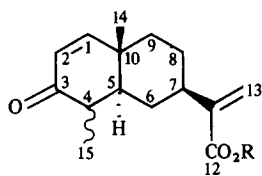
The aerial parts of *Perymenium featherstonei* Blake afforded germacrene D, *ent*-kaurenic acid, its 15 α -angeloyloxy derivative, *ent*-kauren-19-ol, 8-desoxy-

salonitenolide [3], two epimeric keto acids, the eudesmane derivatives **1a** and **2a**, as well as the eudesmanolide **3**. The structure of the latter followed from the molecular formula (C₁₅H₁₈O₃) and the ¹H NMR spectrum (Table 1). Spin decoupling allowed the assignment of all signals though the signals of H-6 and H-9 were overlapped in deuteriochloroform. Addition of deuteriobenzene shifted the signal of H-6 β less than those of H-6 α and H-9 β . Therefore all sequences and couplings could be assigned. The configuration at C-4 followed from the coupling *J*_{4,5}. The couplings of H-8 showed that a *cis*-fused 8,12-lactone was present with the same conformation as alantolactone and related eudesmanolides. The position of the keto group and of the *cis*-double bond directly followed from the ¹H NMR spectrum. All data

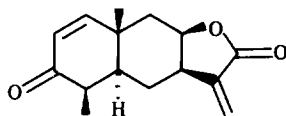
Table 1. ¹H NMR spectral data of compounds **1**–**3** (400 MHz, CDCl₃, TMS as internal standard)

	1a	2a	1b	2b	1b	2b	3	
					CDCl ₃ –C ₆ D ₆			CDCl ₃ –C ₆ D ₆
H-1	6.80 <i>d</i>	6.73 <i>d</i>	6.78 <i>d</i>	6.72 <i>d</i>	6.54 <i>d</i>	6.40 <i>d</i>	6.76 <i>d</i>	6.34 <i>d</i>
H-2	5.89 <i>d</i>	5.87 <i>d</i>	5.87 <i>d</i>	5.85 <i>d</i>	5.81 <i>d</i>	5.75 <i>d</i>	5.89 <i>d</i>	5.71 <i>d</i>
H-4	2.46 <i>dq</i>	2.27 <i>dq</i>	2.45 <i>dq</i>	2.26 <i>dq</i>	2.34 <i>dq</i>	2.06 <i>dq</i>	2.49 <i>dq</i>	2.25 <i>dq</i>
H-5	2.15 <i>ddd</i>	1.72 <i>ddd</i>	2.15 <i>ddd</i>	1.71 <i>ddd</i>	1.95 <i>ddd</i>	1.49 <i>ddd</i>	2.08 <i>ddd</i>	1.63 <i>ddd</i>
H-6 α	1.56 <i>br d</i>	1.8 <i>m</i>	1.51 <i>br d</i>	1.79 <i>br d</i>	1.36 <i>br d</i>	1.62 <i>br d</i>	1.67 <i>m</i>	1.20 <i>m</i>
H-6 β	1.63 <i>ddd</i>	1.33 <i>ddd</i>	1.62 <i>ddd</i>	1.31 <i>ddd</i>	1.59 <i>ddd</i>	1.10 <i>ddd</i>		1.36 <i>ddd</i>
H-7	2.60 <i>dddd</i>	2.51 <i>dddd</i>	2.60 <i>dddd</i>	2.52 <i>dddd</i>	2.50 <i>dddd</i>	2.37 <i>dddd</i>	3.07 <i>dddd</i>	2.52 <i>dddd</i>
H-8 α	1.78 <i>br d</i>	1.8 <i>m</i>	1.74 <i>br d</i>	1.73 <i>br d</i>	1.63 <i>br d</i>	1.56 <i>br d</i>	4.58 <i>ddd</i>	4.11 <i>ddd</i>
H-8 β	1.6 <i>m</i>		1.61 <i>m</i>	1.58 <i>ddd</i>	1.59 <i>ddd</i>	1.39 <i>ddd</i>		
H-9 α	1.51 <i>ddd</i>	1.6 <i>m</i>	1.48 <i>ddd</i>	1.50 <i>ddd</i>	1.36 <i>m</i>	1.39 <i>ddd</i>	2.26 <i>dd</i>	1.88 <i>dd</i>
H-9 β	1.6 <i>m</i>		1.58 <i>m</i>	1.67 <i>m</i>		1.24 <i>m</i>	1.65 <i>dd</i>	1.20 <i>m</i>
H-13	6.36 <i>s</i>	6.36 <i>s</i>	6.19 <i>d</i>	6.18 <i>d</i>	6.13 <i>d</i>	6.07 <i>d</i>	6.20 <i>d</i>	6.01 <i>d</i>
H-13'	5.73 <i>s</i>	5.71 <i>s</i>	5.60 <i>dd</i>	5.58 <i>dd</i>	5.44 <i>dd</i>	5.36 <i>dd</i>	5.65 <i>d</i>	5.29 <i>d</i>
H-14	1.20 <i>s</i>	1.11 <i>s</i>	1.19 <i>s</i>	1.10 <i>s</i>	1.02 <i>s</i>	0.86 <i>s</i>	1.25 <i>s</i>	0.97 <i>s</i>
H-15	1.13 <i>d</i>	1.12 <i>d</i>	1.19 <i>d</i>	1.11 <i>d</i>	1.03 <i>d</i>	1.01 <i>d</i>	1.12 <i>d</i>	0.90 <i>d</i>
OMe	—	—	3.77 <i>s</i>	3.77 <i>s</i>	3.63 <i>s</i>	3.55 <i>s</i>	—	—

J (Hz): Compounds **1a**/**1b** and **2a**/**2b**: 1, 2 = 10; 5, 6 α = 3; 5, 6 β = 12; 6 α , 6 β = 13; 6 α , 7 = 4; 6 β , 7 = 12; 7, 8 α ~ 3; 7, 8 β = 12; 8 α , 8 β = 13; 7, 13' = 1; 8 α , 9 α ~ 3; 8 α , 9 β ~ 3; 8 β , 9 α ~ 10; 8 β , 9 β ~ 5; 9 α , 9 β = 13; 13, 13' = 1; compounds **1a**/**1b**: 4, 5 = 6; 4, 15 = 8; compounds **2a**/**2b**: 4, 5 = 12.5; 4, 15 = 7; compound **3**: 1, 2 = 13; 4, 5 = 6; 4, 15 = 8; 5, 6 α = 2.5; 5, 6 β = 13; 6 α , 6 β = 13; 6 α , 7 ~ 6; 6 β , 7 = 11.5; 7, 8 = 5; 7, 13 = 1; 7, 13' = 1; 8, 9 α = 5; 8, 9 β = 1.5; 9 α , 9 β = 15.



- 1a** R = H, 4 α H
1b R = Me, 4 α H
2a R = H, 4 β H
2b R = Me, 4 β H



3

therefore only agree with the structure of 4 α ,15-dihydroencelin (3).

The structures of the acids **1a** and **2a**, which were transformed to the corresponding methyl esters **1b** and **2b**, also followed from the molecular formulae and the ^1H NMR spectral data (Table 1). All signals in the ^1H NMR spectra of **1b** could be assigned by spin decoupling starting with the five fold doublet at δ 2.60, clearly the signal of H-7 which followed from decoupling of the H-13' signal and of four other signals. The stereochemistry at C-4 again could be deduced from the corresponding coupling $J_{4,5}$ and that at C-7 was deduced from the presence of two large couplings ($J_{6,7\alpha}$ and $J_{7\alpha,8\beta}$).

The ^1H NMR spectral data of **2b** differed only slightly from those of **1b**. A clear difference, however, was detectable between the chemical shifts of some protons. The coupling $J_{4,5}$ in the spectrum of **2b** required a *trans*-axial orientation of the protons at C-4 and C-5 thus indicating that **1a** and **2a** most likely were C-4 epimers which could be established by complete epimerization of **1a** to **2a**. The observed shift differences in the spectra of **1b** and **2b** probably required different conformations of the cyclohexenone ring in both epimers. Most likely **1b** adopts a boat-conformation to avoid a 1,3-diaxial orientation of the methyl groups at C-4 and C-10, while for **2b** a normal half chair conformation can be assumed.

The co-occurrence of **1a** and **3** may be an indication that 8,12-*cis*-eudesmanolides may be formed via the corresponding eudesmanoic acids rather than from 8,12-*cis*-germacranolides which seem to be rare.

The roots afforded the widespread trideca-3,5,7,9,11-pentayn-1-ene, germacrene D, *ent*-kaurenic acid, its 15 α -angeloyloxy and isobutyryloxy derivatives and *ent*-kaurenol.

The isolation of the eudesmanolide **3** from a *Perymenium* species supports the placement of this genus in the Euphorbiaceae where eudesmanolides and *ent*-kaurenic acid derivatives are common (*Aspilula* [4], *Baltimora* [5],

Dimerostemma [6], *Encelia* [7–9], *Steiractinia* [4], *Wedelia* [10], *Zexmenia* [11], *Zinnia* [12]). From a few genera cotic acid derivatives were reported which may replace the eudesmanolides [13–15]. However, several other types of natural product are present in this somewhat diverse subtribe.

EXPERIMENTAL

The air dried plant material (voucher RMK 9025) was extracted with Et₂O–petrol, 1:2, and the resulting extracts were evaporated under vacuum. The extract of the aerial parts (300 g) gave CC fractions (silica gel) as follows: 1 (100 ml, petrol), 2 (200 ml, Et₂O–petrol, 1:10 and 1:3), 3 (100 ml, Et₂O–petrol, 1:1), 4 (200 ml, Et₂O and Et₂O–MeOH, 10:1). TLC (silica gel, PF 254, petrol) of fraction 1 gave 5 mg germacrene D (R_f 0.60; detection always by UV-light and KMnO₄ spray). TLC of 2 (Et₂O–petrol, 1:10) afforded 1 g *ent*-kaurenic acid (R_f 0.45), TLC of 3 (Et₂O–petrol, 1:1) gave 50 mg 15 α -angeloyloxy-*ent*-kaurenic acid (R_f 0.63) and 100 mg *ent*-kauren-19-ol (R_f 0.38). Fraction 4 on repeated TLC (Et₂O–petrol, 3:1) afforded 10 mg **2a** (R_f 0.65), 20 mg **1a** (R_f 0.61) and a mixture of more polar compounds, which after TLC (C₆H₆–Et₂O–CH₂Cl₂, 1:1:1) gave 3 mg **3** (R_f 0.63) and ca 2 mg crude 8-desoxysalonitenolide which was purified by HPLC (RP 8, MeOH–H₂O 13:7, R_f 6:1 min.).

The extract of the roots (100 g) gave fractions as follows: 1 (50 ml, petrol), 2 (100 ml, Et₂O–petrol, 1:10 and 1:3), 3 (100 ml, Et₂O–petrol, 1:1) and 4 (100 ml Et₂O and Et₂O–MeOH, 10:1). TLC (silica gel PF 254) of fraction 1 (petrol) gave 5 mg germacrene D and ca 0.5 mg tridecapentayn-1-ene (estimated by UV-extinction). TLC of 2 (Et₂O–petrol, 1:10) gave 0.5 g *ent*-kaurenic acid (R_f 0.48), and TLC of 3 (Et₂O–petrol, 1:1) afforded 20 mg 15 α -angeloyloxy- and 20 mg 15 α -isobutyryloxy-*ent*-kaurenic acid (R_f 0.60) while fraction 4 gave no definite compounds. Known compounds were identified by comparing their 400 MHz ^1H NMR spectra with those of authentic material. Quantities were estimated by weight.

3-Oxo-1,2-dehydro-4 α ,15-dihydro cotic acid (1a). Colourless oil; MS m/z (rel. int.): 248.141 [M]⁺ (44) (C₁₅H₂₀O₃), 233 [M –Me]⁺ (8), 230 [M –H₂O]⁺ (16), 202 [230–CO]⁺ (14), 95 [C₇H₁₁]⁺ (90), 67 [C₅H₇]⁺ (100). Addition of CH₂N₂ in Et₂O afforded the methyl ester **1b** (TLC, Et₂O–petrol, 1:3 R_f 0.48), colourless oil, bp_{0.1} 125° (bath temp. short way dest.); IR $\nu_{\text{max}}^{\text{CCl}_4}$ cm^{–1}: 1720 (C=CCO₂R), 1675, 1620 (C=CC=O); MS m/z (rel. int.): 262.157 [M]⁺ (100) (C₁₆H₂₂O₃), 247 [M –Me]⁺ (12), 230 [M –MeOH]⁺ (44), 215 [230–Me]⁺ (34), 202 [230–CO]⁺ (56), 95 (79), 91 (89), 67 (100).

$$[\alpha]_{24}^{25} = \frac{589}{-25} \frac{578}{-26} \frac{546}{-36} \frac{436}{-152} \text{nm} \text{CHCl}_3; c \text{ 1.34}$$

To 10 mg **1a** in 1 ml tert.-butanol 5 mg potassium-tertbutylate was added. After 1 h, usual work-up afforded **2a**, identical with the natural acid, while nothing of the epimer was detectable.

3-Oxo-1,2-dehydro-4 β ,15-dihydro cotic acid (2a). Colourless oil; MS m/z (rel. int.): 248.141 [M]⁺ (36) (C₁₅H₂₀O₃), 233 [M –Me]⁺ (8), 230 [M –H₂O]⁺ (9), 202 [230–CO]⁺ (14), 95 (50), 67 (58), 61 (100). Addition of CH₂N₂ in Et₂O gave the methyl ester **2b** (TLC, Et₂O–petrol, 1:3 R_f 0.45), colourless oil, bp_{0.1} 125° (bath temp., short way dest.); IR $\nu_{\text{max}}^{\text{CCl}_4}$ cm^{–1}: 1720 (C=CCO₂R), 1675, 1620 (C=CC=O); MS m/z (rel. int.): 262.157 [M]⁺ (61) (C₁₆H₂₂O₃), 247 [M –Me]⁺ (8), 230 [M –MeOH]⁺ (41), 215 [230–Me]⁺ (16), 202 [230–CO]⁺ (31), 95 (77), 91 (78), 67 (100).

$$[\alpha]_{24}^{25} = \frac{589}{-50} \frac{578}{-52} \frac{546}{-61} \frac{436}{-130} \text{nm} \text{CHCl}_3; c \text{ 0.63}$$

4 α ,15-Dihydroencelin (3). Colourless crystals, mp 175° (Et₂O); IR $\nu_{\text{max}}^{\text{CCl}_4}$ cm⁻¹: 1760 (γ -lactone), 1670 (C=CC=O); MS m/z (rel. int.): 246.126 [M]⁺ (21) (C₁₅H₁₈O₃), 231 [M - Me]⁺ (3), 218 [M - CO]⁺ (3), 97 (71), 95 (79), 69 (100), 55 (98).

$$[\alpha]_{24}^{\lambda} = \frac{589}{+14} \frac{578}{+14} \frac{546}{+20} \frac{436 \text{ nm}}{+53} \text{CHCl}_3; c \text{ 0.1}$$

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2-ACETOXY-3 α ,4 α -EPOXY-3,4-DIHYDROKAUNIOLIDE FROM *GROSVENORIA COELOCAULIS*

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Key Word Index—*Grosvenoria coelocaulis*; Compositae; sesquiterpene lactones; guaianolides.

Abstract—The aerial parts of *Grosvenoria coelocaulis* gave the known guaianolides dehydroleucodin, desacetoxymatricarin, kauniolide and a new one, 2 β -acetoxy-3 α ,4 α -epoxy-3,4-dihydrokauniolide.

Grosvenoria (Compositae, tribe Eupatorieae) is a small genus [1] ranging from central Ecuador into northern Peru, which is placed in the subtribe Critoniinae [2]. So far nothing is known on the chemistry of this genus. We have now investigated *Grosvenoria coelocaulis* (B. L. Robins.) K. et R. from northern Peru. The aerial parts afforded α -curcumene and α -zingiberene as well as the known guaianolides kauniolide (1) [3], dehydroleucodin [4], desacetoxymatricarin [5] together with a new one, C₁₇H₂₀O₅. The IR spectrum of the latter indicated the presence of a γ -lactone and an acetate group (1770, 1738, 1245 cm⁻¹). From the ¹H NMR spectrum (Table 1) the presence of 6 α ,12-methylene lactone could be deduced. A typical fourfold doublet at δ 2.80 was due to the H-7 signal. Accordingly, its irradiation collapsed the H-13 doublets to singlets, the double doublet at δ 3.63 to a doublet and changed the overlapped multiplet around

2.10 while a quartet at 1.33 collapsed to a triplet. Addition of deuteriobenzene allowed the assignment of all signals by spin decoupling. As H-2 showed a *W*-coupling with H-5 and an allylic coupling with H-14 the whole sequence leading to the structure **2** could be assigned. The chemical shifts of H-2 and H-3 in combination with the other data indicated a 2-acetoxy derivative of kauniolide where the 3,4-double bond was transformed to an epoxide. Inspection of a model showed that the small coupling $J_{2,3}$ required a *trans*-orientation of H-2 and H-3, while the chemical shift of H-15 obviously was influenced by the deshielding effect of the lactone oxygen at C-6. This, however, required a 4 β -methyl group as the stereochemistry at C-5–C-7 clearly followed from the *trans*-diaxial couplings of H-5–H-7. Also the observed couplings of H-8 and H-9 nicely agreed with the angles which could be deduced from a model. Thus the new compound is 2 β -